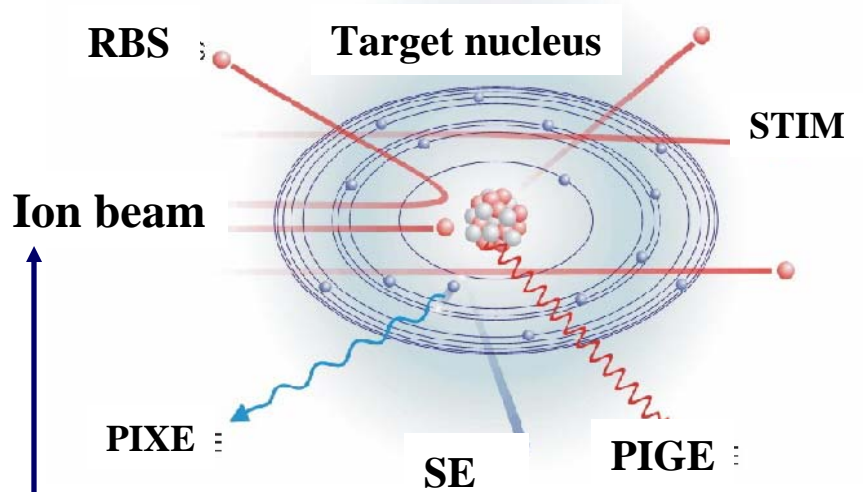


Ion-beam techniques



- RBS:** Rutherford backscattering
- ERD:** Elastic recoil detection
- PIXE:** Particle induced x-ray emission
- PIGE:** Particle induced gamma emission
- NRA:** Nuclear reaction analysis
- STIM:** Scanning Transm. Ion Microscopy
- SE:** Secondary emission
- CH:** Channeling

Electrostatic Accelerators

Van de Graaff accelerator
 Pelletron
 Tandem Van de Graaff

TECHNIQUE	ION BEAM	ENERGY (MeV)
PIXE	H ⁺	1 - 4
RBS	⁴ He ⁺ , H ⁺	≤ 2
ERD	³⁵ Cl ⁺ , ²⁰ Ne ⁺ ³ He ⁺ , ⁴ He ⁺	2 - 40
NRA	H ⁺ , D ⁺	0.4 - 3

Some literature on ion beam analysis

PIXE: A Novel Technique for Elemental Analysis

Sven A. E. Johansson and John L. Campbell

Publisher: John Wiley & Sons, 1988

Materials Analysis using a Nuclear Microprobe

M B H Breese, D N Jamieson and P J C King

Publisher: John Wiley & Sons, 1996

Handbook of Modern Ion Beam Materials Analysis

Edited by Joseph R. Tesmer and Michael Nastasi

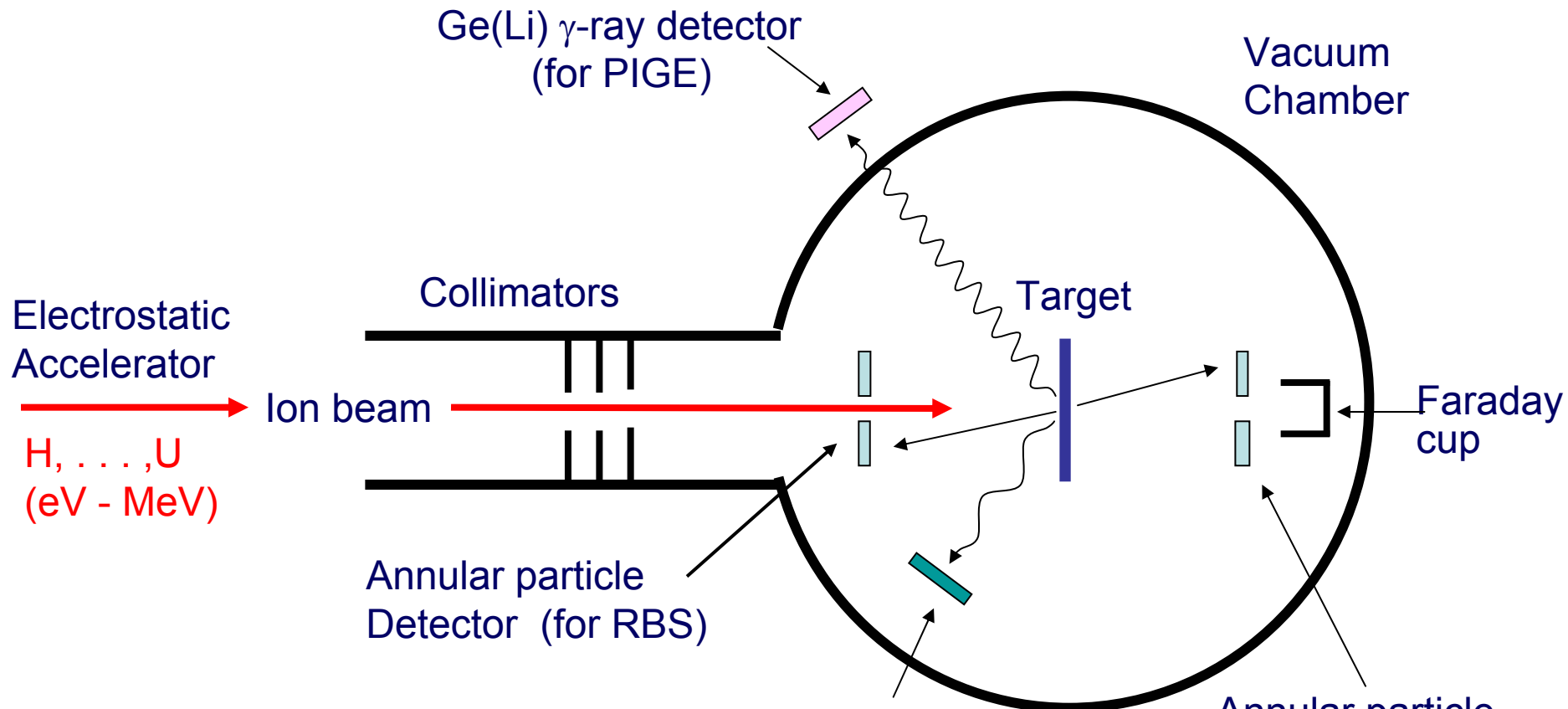
Publisher: Materials Research Society, Pittsburgh, Pa., 1995

Handbook of X-Ray Spectrometry

Edited by Rene E. Van Grieken and Andrzej A. Markowicz Publisher:

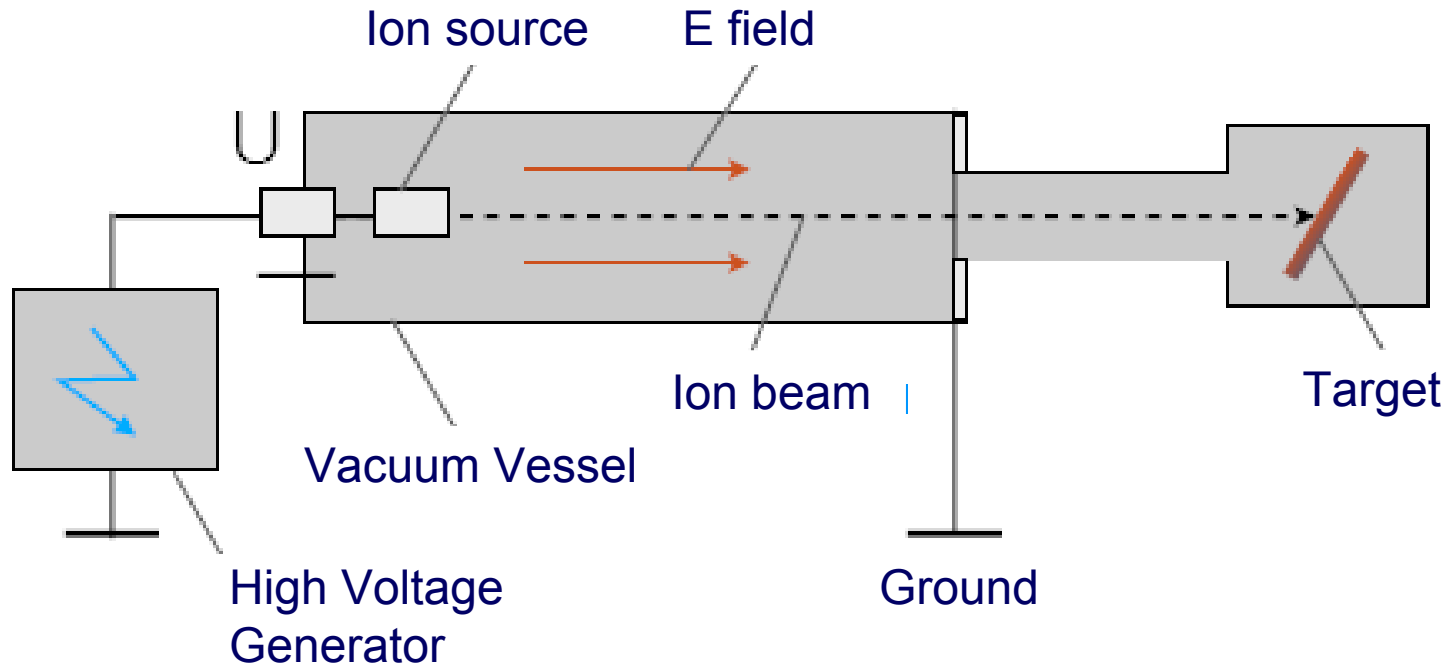
Marcel Dekker, Inc., 2nd Edition, 2002

Experimental Setup for Ion Beam Analysis



Detectors:	ΔE
Si (Li)	120 – 200 eV
Ge (high purity)	120 – 600 eV
Si (drift chamber)	150 – 200 eV
Crystal spectrometer	1 eV

Electrostatic Accelerator



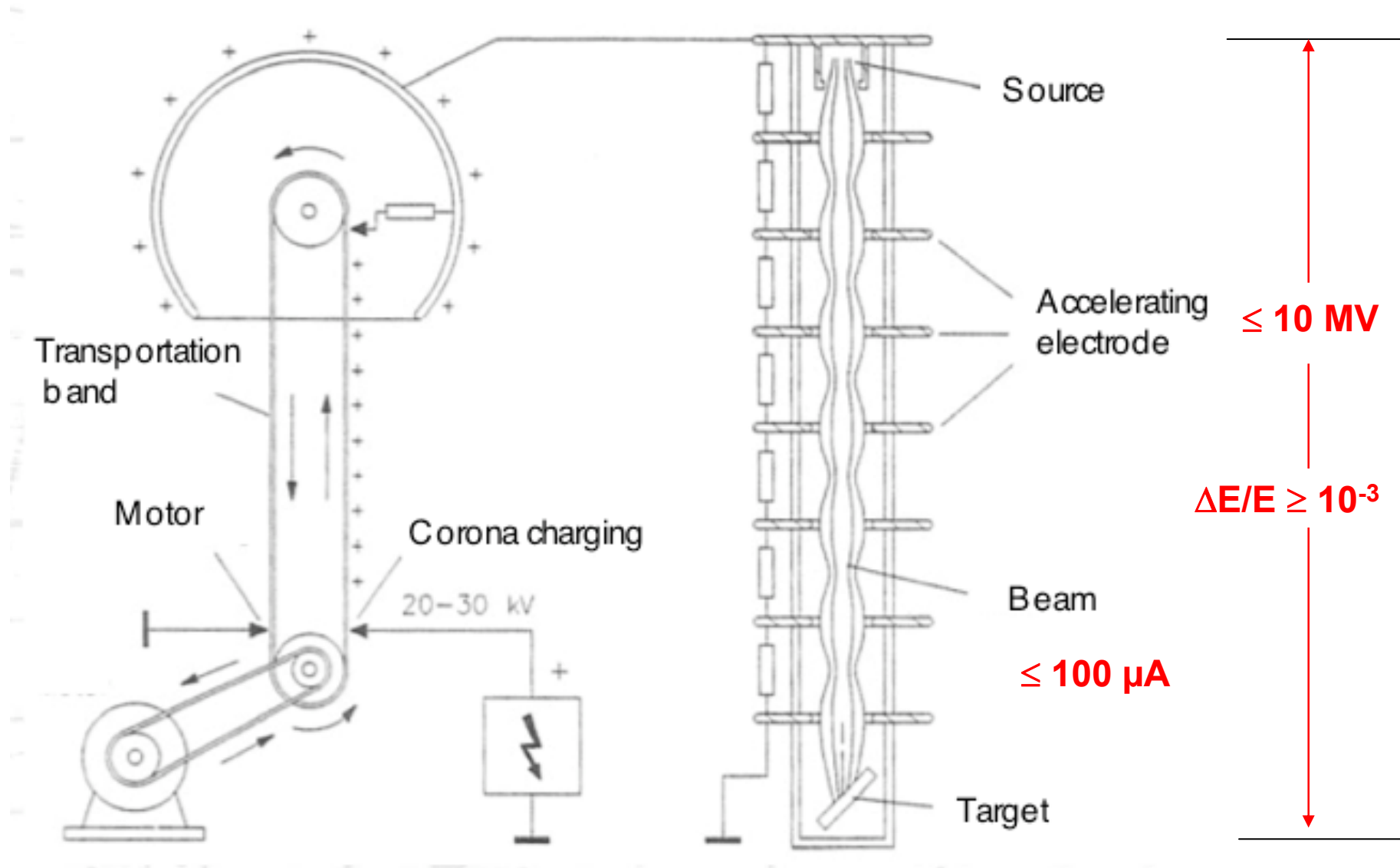
Kinetic particle energy: $E_{\text{kin}} = q U = Ze U$

Van de Graaff: $U \leq 10 \text{ MV}$

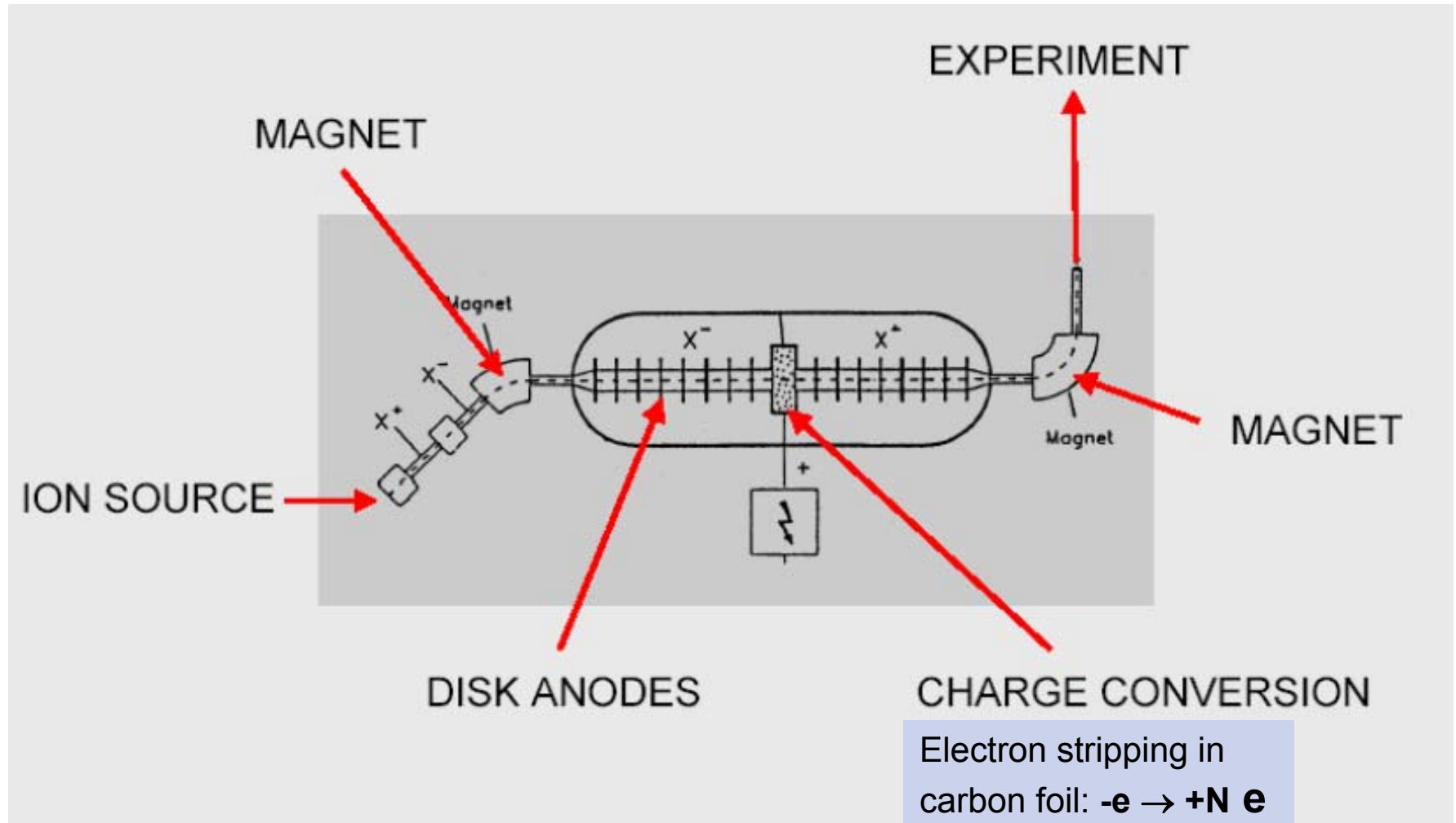
Tandem: $U \leq 20 \text{ MV}$

Tandem Pelletron: $U \leq 30 \text{ MV}$

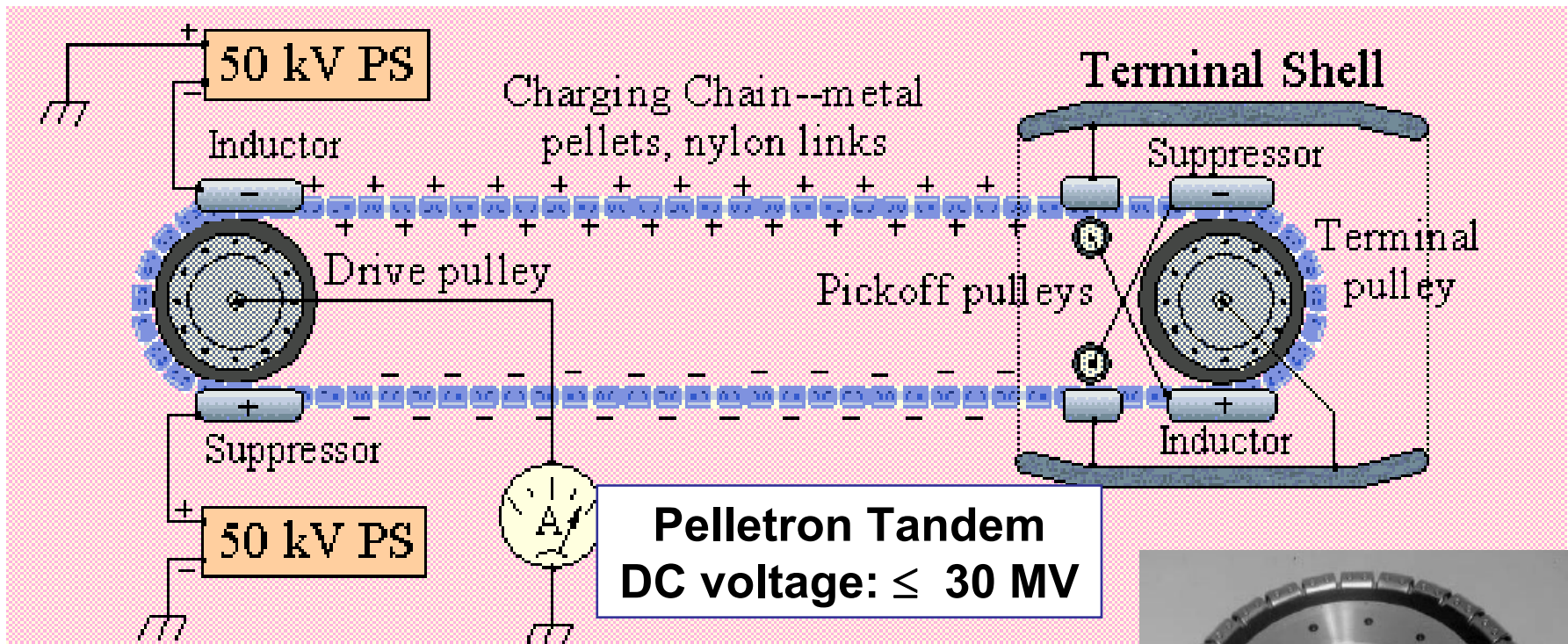
Van de Graaff accelerator



Tandem van de Graaff accelerator

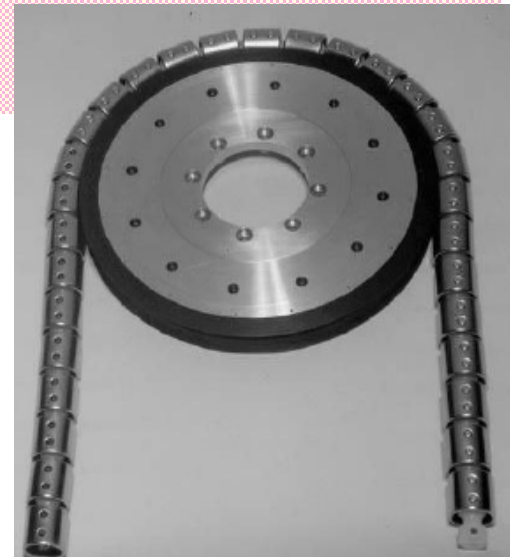


Pelletron charging system



Pelletron

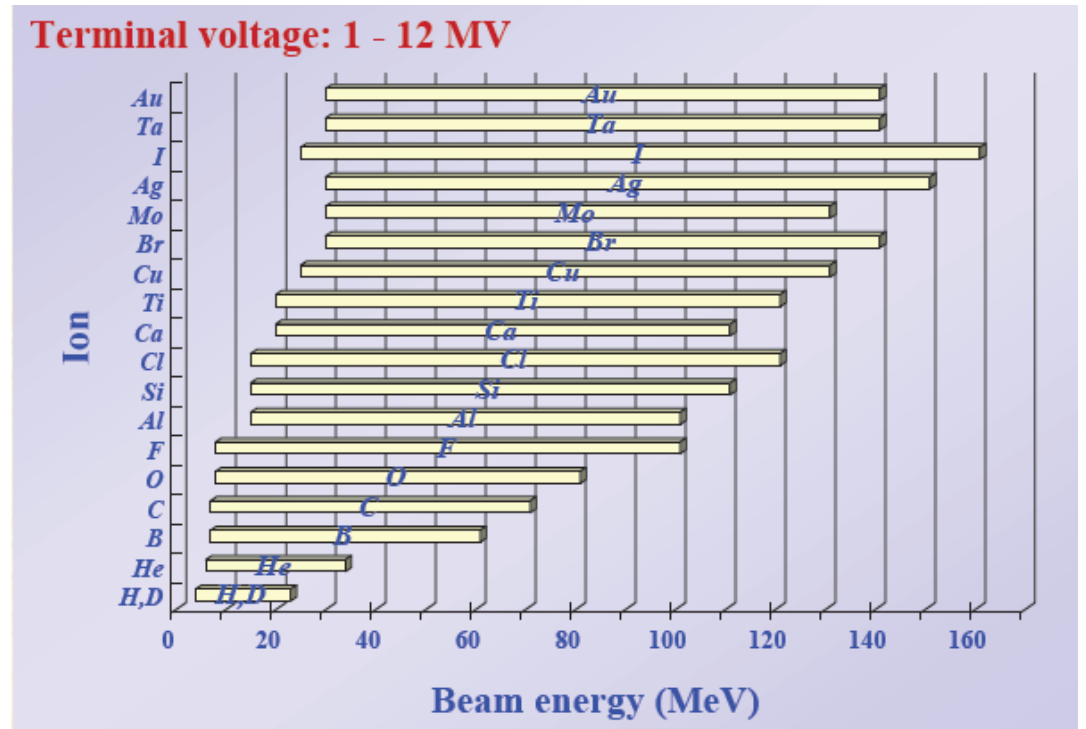
Rubber belt is replaced by a chain of short conductive rods connected by insulating links. Can be operated at much higher velocity than a belt, so both the voltage and currents attainable are higher than a conventional Van de Graaff machine.



USP 8UD Pelletron tandem accelerator



Beam energies available at Tsukuba 12UD Pelletron tandem

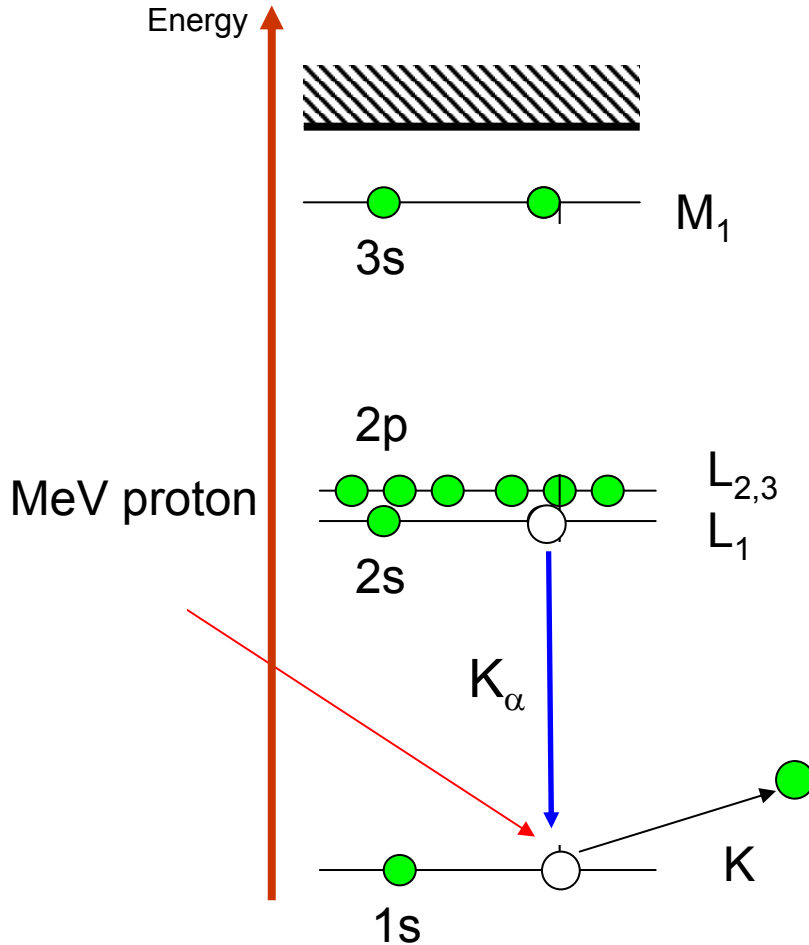


TYPICAL ION BEAMS AND INCIDENT ENERGIES USED IN VARIOUS ION BEAM TECHNIQUES

TECHNIQUE	ION BEAM	ENERGY (MeV)	REMARK
PIXE	H ⁺	1 - 4	Maximum sensitivity in atomic ranges 10<Z<35 and 75<Z<85
RBS	⁴ He ⁺ , H ⁺	≤ 2	Non-Rutherford scattering becomes significant for energy >2 MeV
ERDA	³⁵ Cl ⁺ , ²⁰ Ne ⁺ ³ He ⁺ , ⁴ He ⁺	2 - 40	Mass of incident ion must be greater than that of target nucleus. ³ He ⁺ and ⁴ He ⁺ are used only for the measurement of H.
NRA	H ⁺ , D ⁺	0.4 - 3	Reactions used include (p,g) (p,p'g), (p,ag), (d,p), (d,pg)

Proton Induced X-ray Emission = PIXE

Lund Institute of Technology - 1970



Bohr:

$$E_n = -\frac{me^4 Z^2}{8\epsilon_0^2 h^2 n^2} = -\frac{e^2}{8\pi\epsilon_0 a_0} \frac{Z^2}{n^2} \approx -13,6 \frac{Z^2}{n^2} \text{ eV.}$$

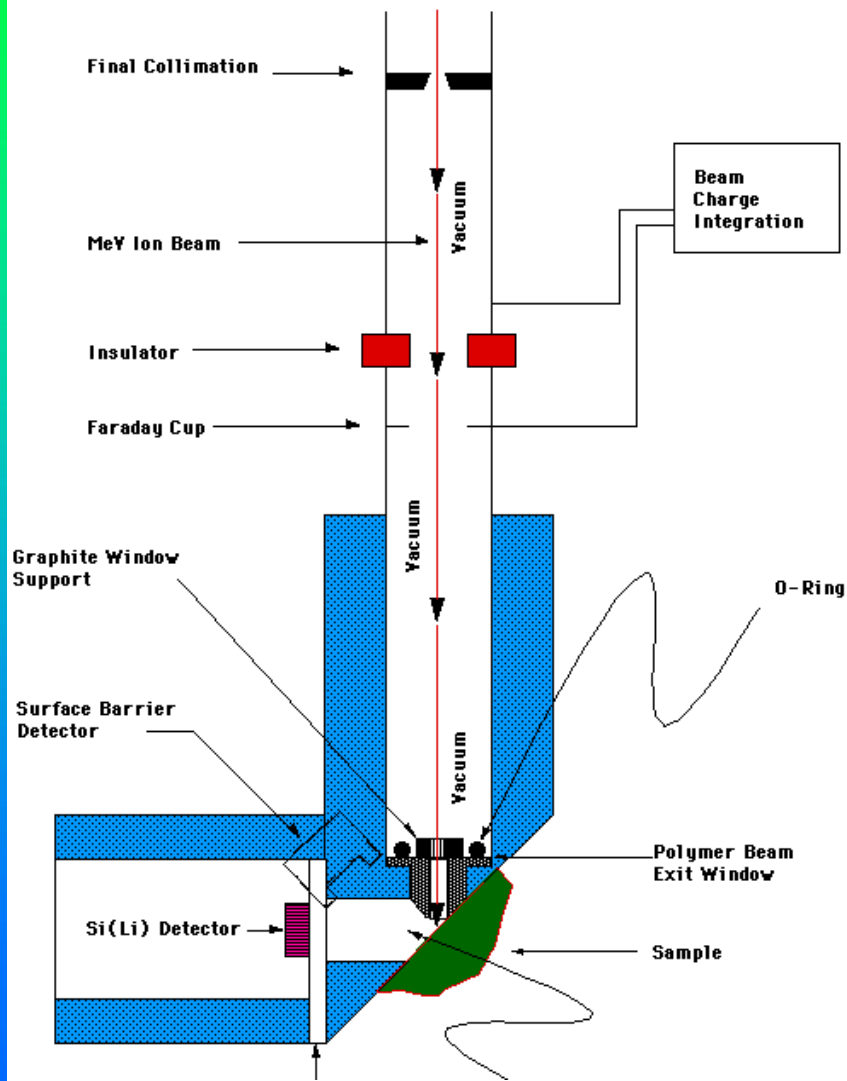
$$\Delta E = E_{n_2} - E_{n_1} = \frac{me^4 Z^2}{8\epsilon_0^2 h^2} \left(\frac{1}{n_1^2} - \frac{1}{n_2^2} \right),$$

K_{α} -line

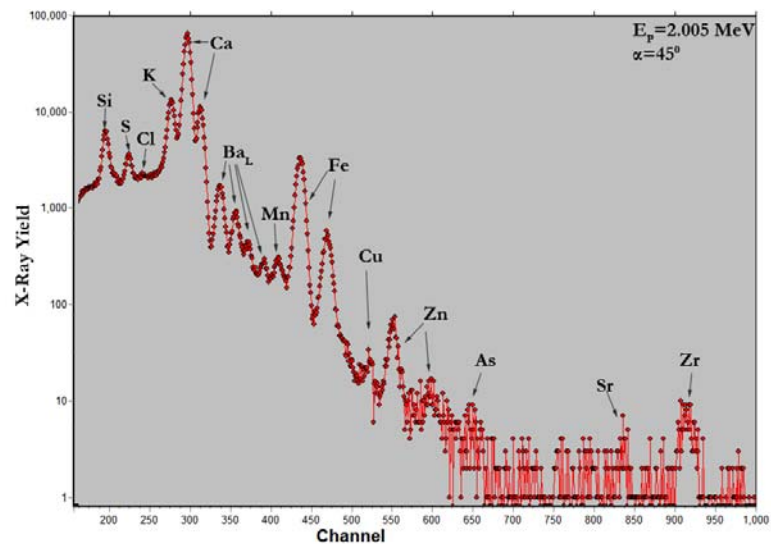
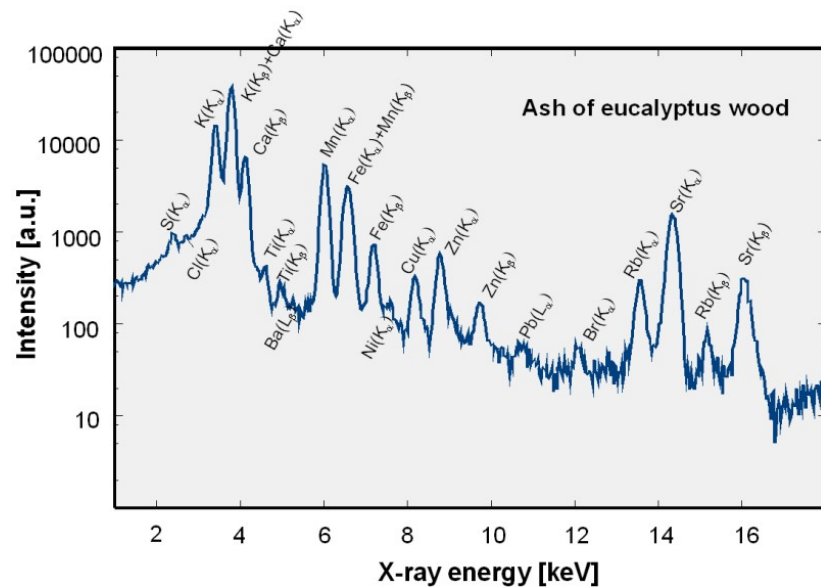
$$h\nu = (2,48 * 10^{15} \text{ Hz})(Z - 1)^2$$

Z		E(K_{α}) (keV)
8	O	0.5249
20	Ca	3.69168
40	Zr	15.7751
70	Yb	52.3889
92	U	98.439

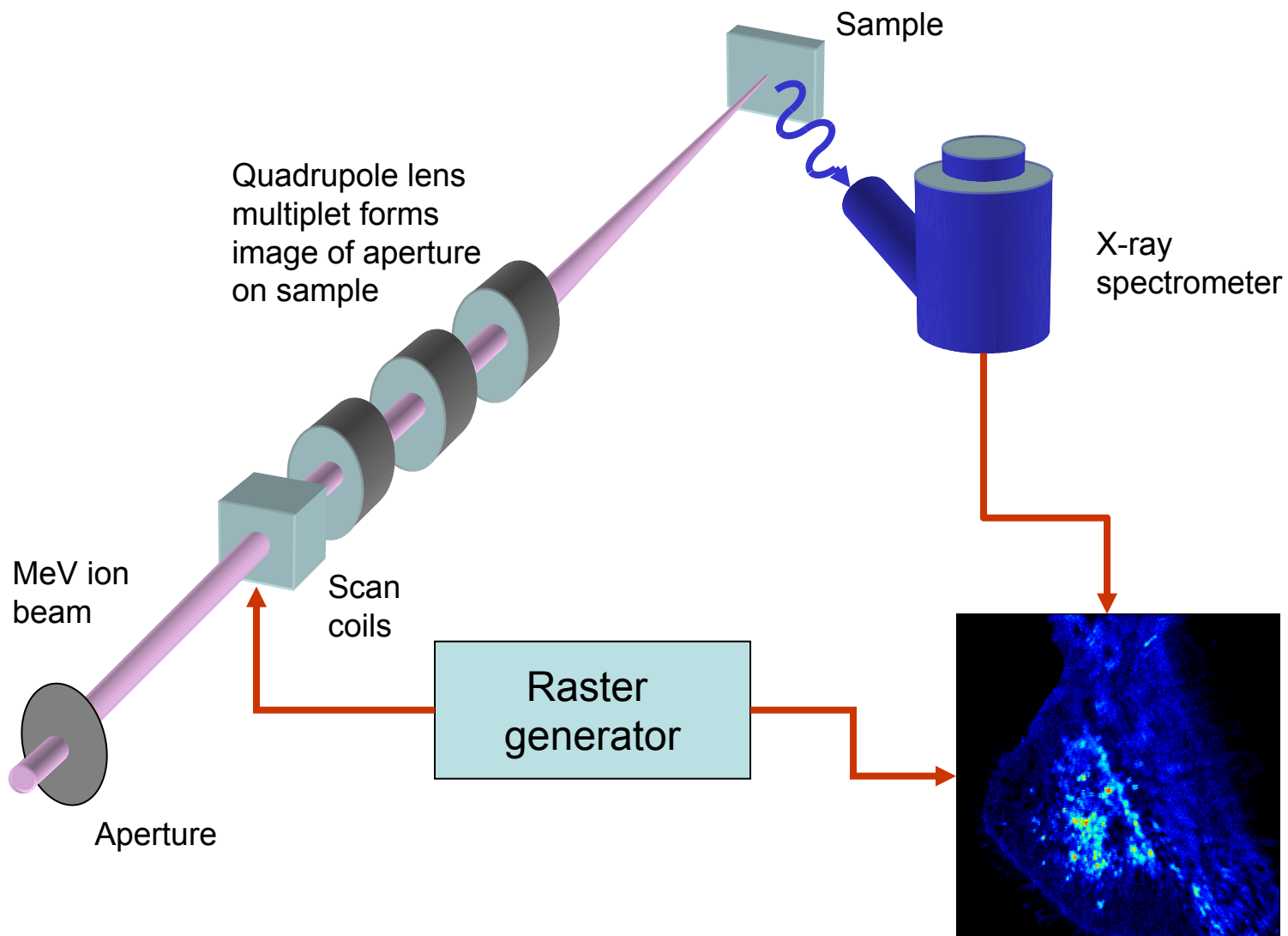
PIXE System (Harvard)



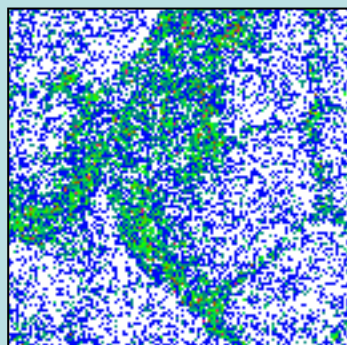
PIXE Spectra



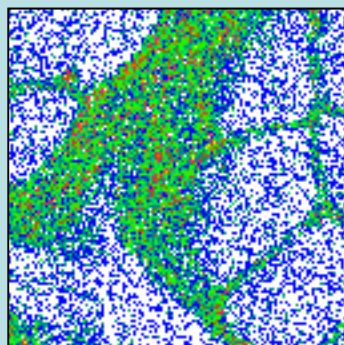
PIXE microscope



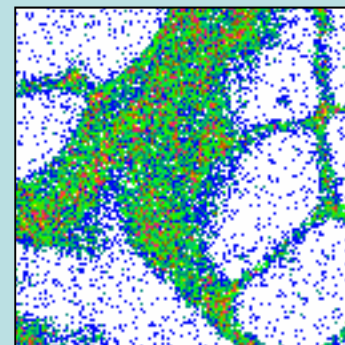
Nuclear microprobe PIXE elemental maps from 400 mm x 400 mm scan over a section of a lung tissue taken from a patient suffered from hard metal lung disease:



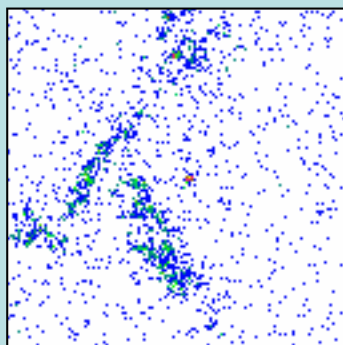
P



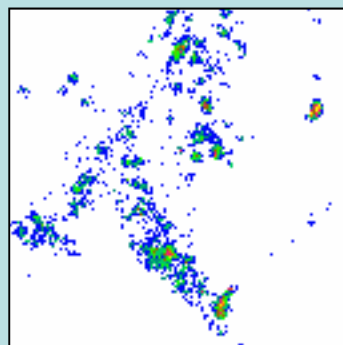
S



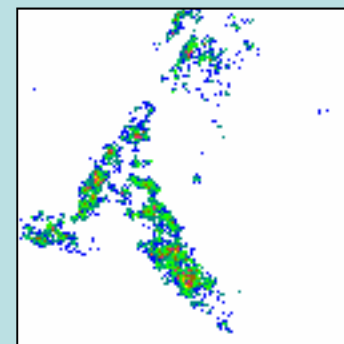
Ca



Ti



Fe



W

PIXE features

All elements with $Z > 14$ (Si) ($K\alpha = 1.73998$ KeV) can be analyzed. For instance in biological samples the concentrations of about 15 elements are normally determined **simultaneously**.

The method is **fast**. A typical irradiation lasts about **10 min**.

Highly sensitive: Concentrations of elements down to **a few 100 ppb (10^{-8})** can be measured.

The spatial resolution of the beam is 1 mm. The penetration depth in a solids: **$\sim 100 \mu\text{m}$** ; allows determinations of **elemental distributions or profiles**

High accuracy: relative errors are between 1-10 %

Small-size samples \geq few mg

Non-destructive technique: This enables analyses of precious objects, e.g. ancient coins or paintings.

Competition to PIXE: Neutron Activation Analysis (NAA)

Neutron sources

Nuclear reactors

Nuclear reactions

$3\text{H}(p,n)3\text{He}$, $6\text{Li}(p,n)6\text{Be}$,

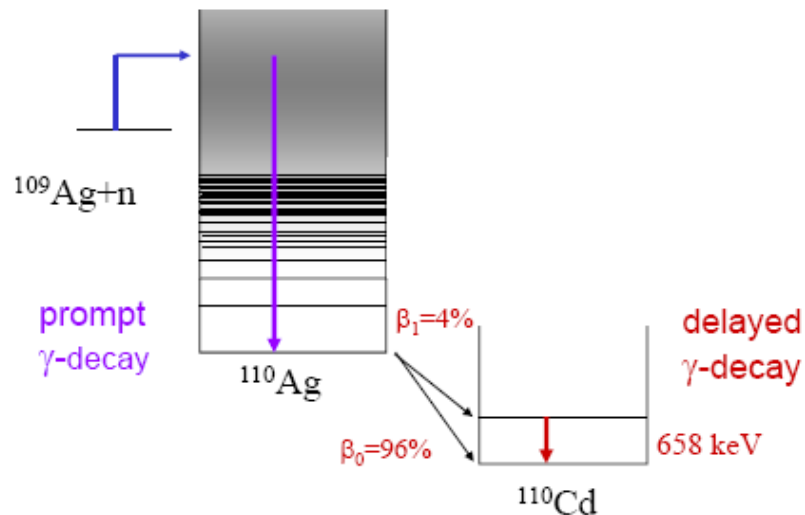
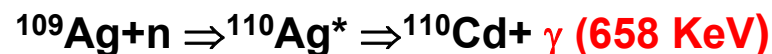
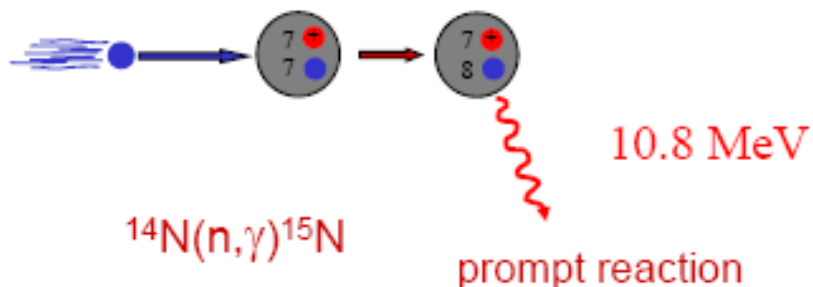
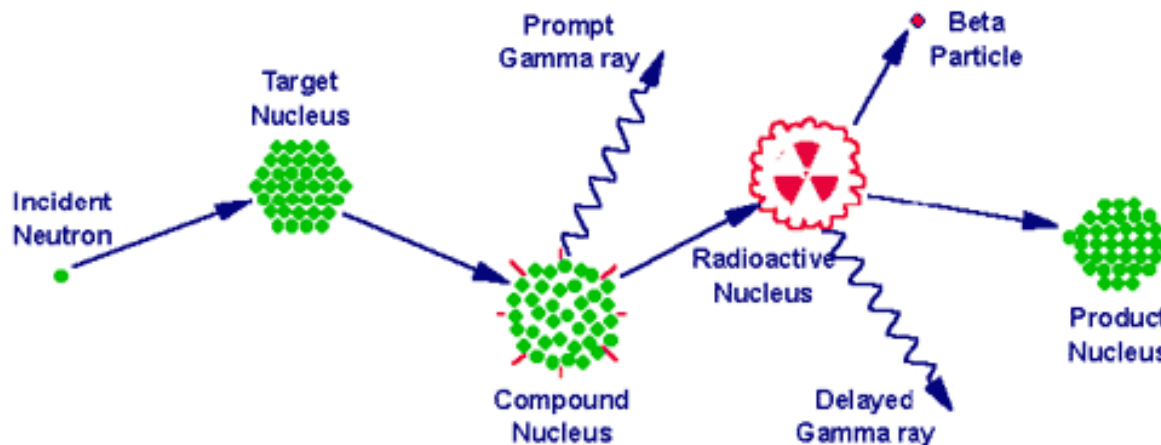
$7\text{Li}(p,n)7\text{Be}$, $9\text{Be}(p,n)9\text{B}$,

Radioisotopes

Spontaneous fission **Cf-252**

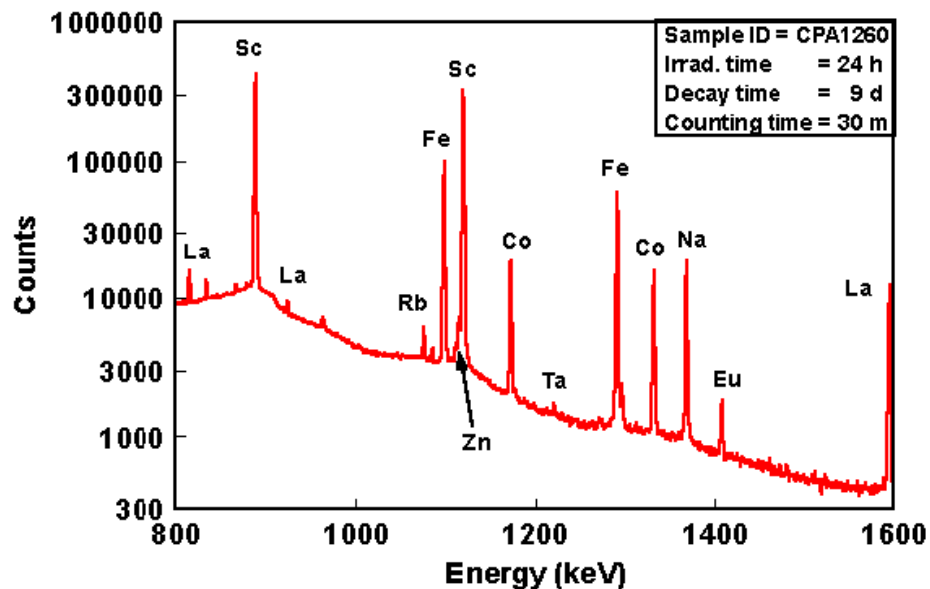
(α,n) reactions

e.g. $^{241}\text{Am}:\text{Be}$



Neutron Activation Analysis

Gamma spectrum

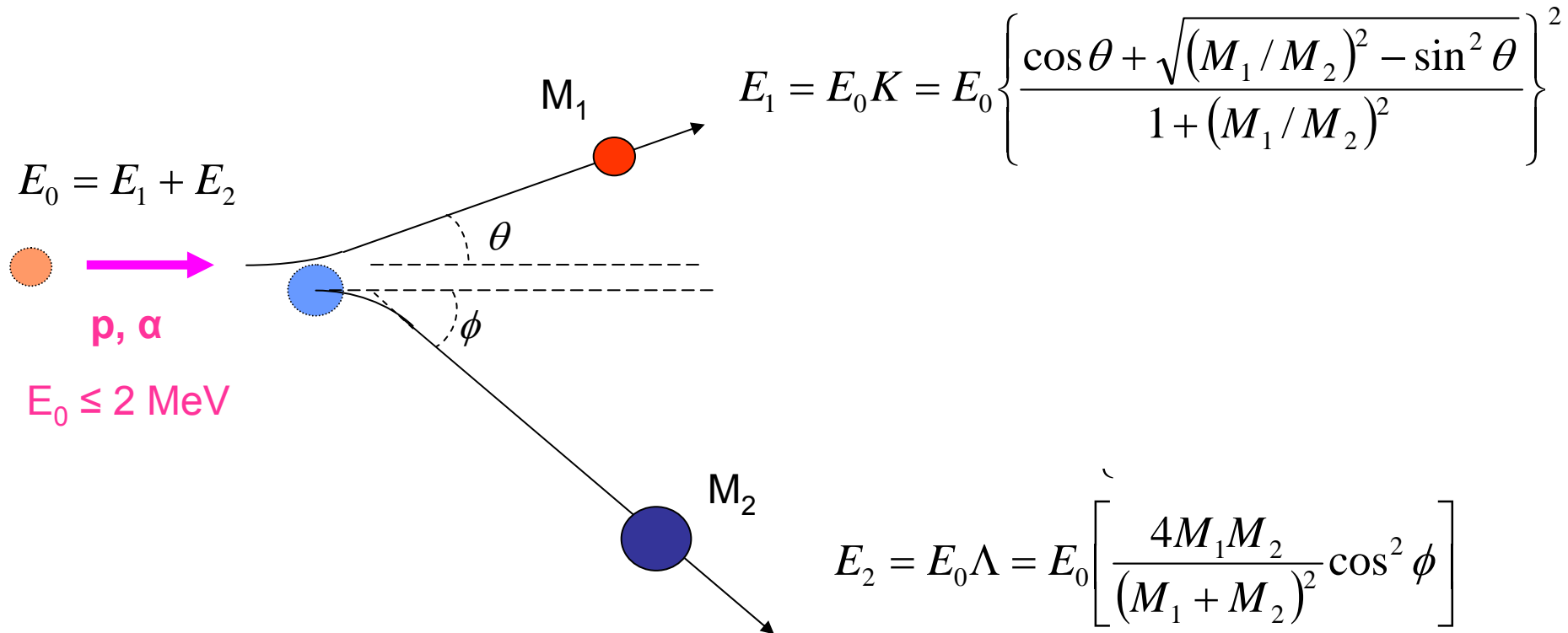


Sensitivity (picograms)	Elements
1	Dy, Eu
1–10	In, Lu, Mn
10–100	Au, Ho, Ir, Re, Sm, W
100–1E3	Ag, Ar, As, Br, Cl, Co, Cs, Cu, Er, Ga, Hf, I, La, Sb, Sc, Se, Ta, Tb, Th, Tm, U, V, Yb
1E3–1E4	Al, Ba, Cd, Ce, Cr, Hg, Kr, Gd, Ge, Mo, Na, Nd, Ni, Os, Pd, Rb, Rh, Ru, Sr, Te, Zn, Zr
1E4–1E5	Bi, Ca, K, Mg, P, Pt, Si, Sn, Ti, Tl, Xe, Y
1E5–1E6	F, Fe, Nb, Ne
1E7	Pb, S

RBS - Rutherford Backscattering Spectrometry

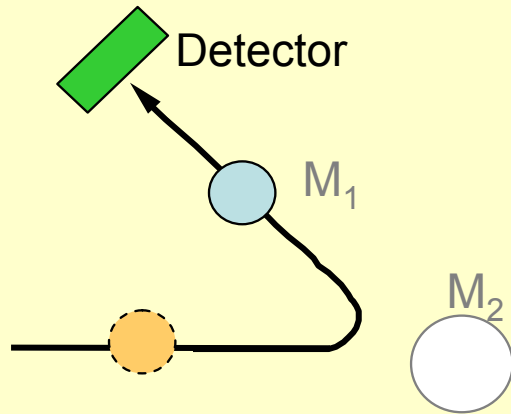
ERD – Elastic recoil detection

Coulomb repulsion between high energy incident ions and target nuclei

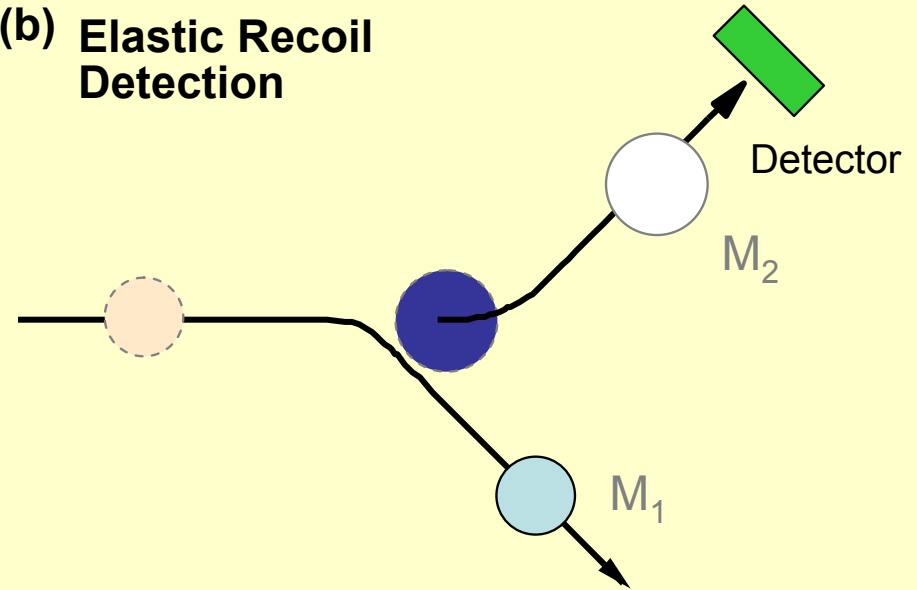


RBS vs. ERDA

(a) Backscattering Spectrometry



(b) Elastic Recoil Detection



- **RBS**

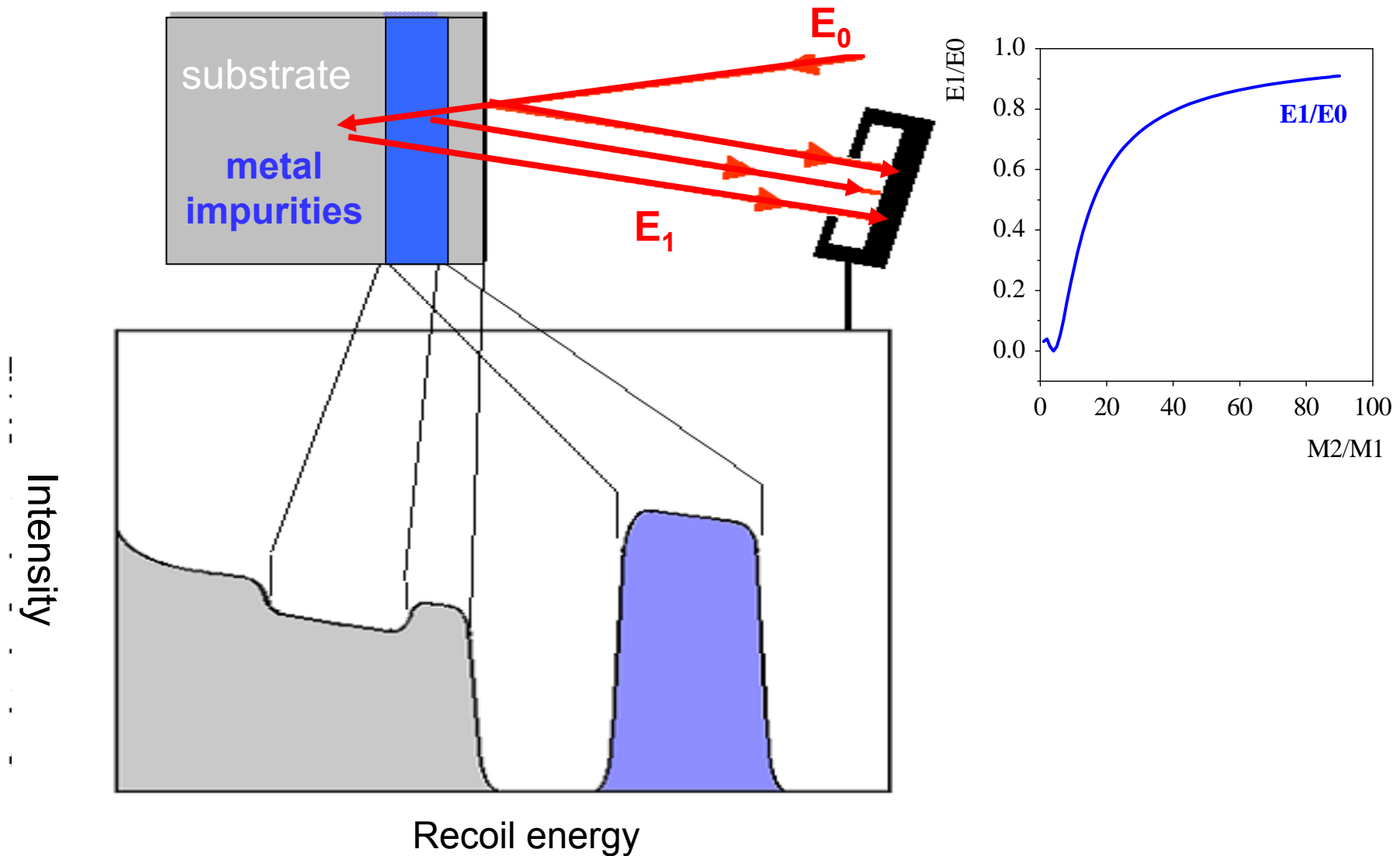
- Projectile is the detected particle $M_1 \ll M_2$
- Depth and mass information encoded on detected particle energy

- **ERDA**

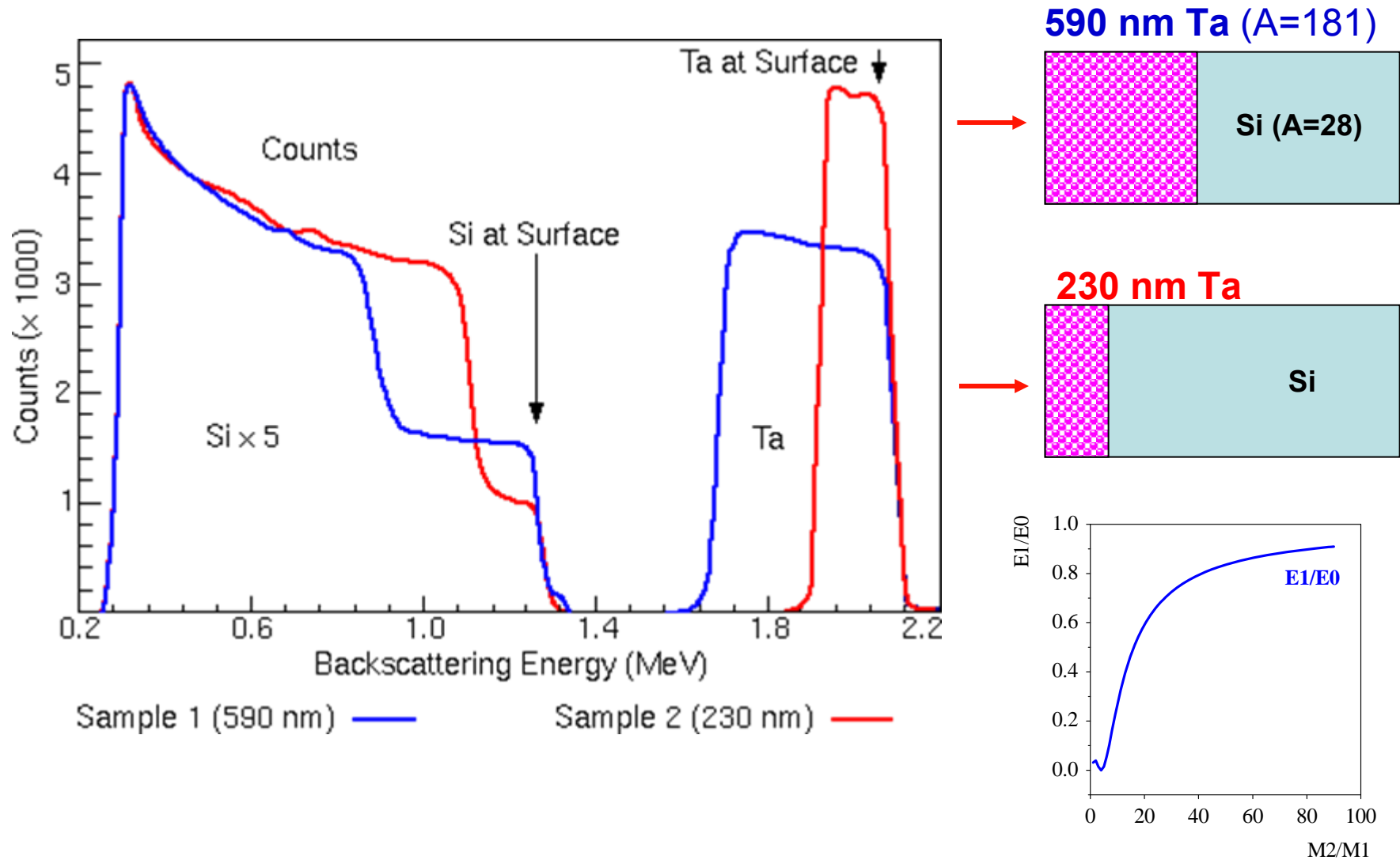
- The target atom is the detected particle $M_1 \gg M_2$
- Depth information encoded on energy
- Isotopic identity carried directly by recoil

RBS (Rutherford Backscattering Spectrometry)

The recoil energy depends on the target mass and the path length



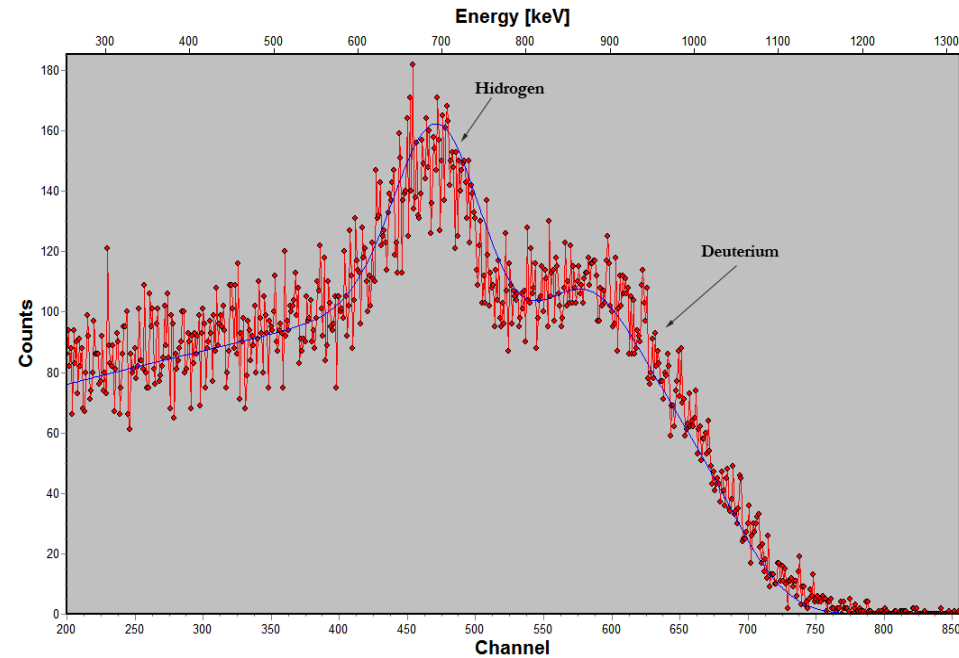
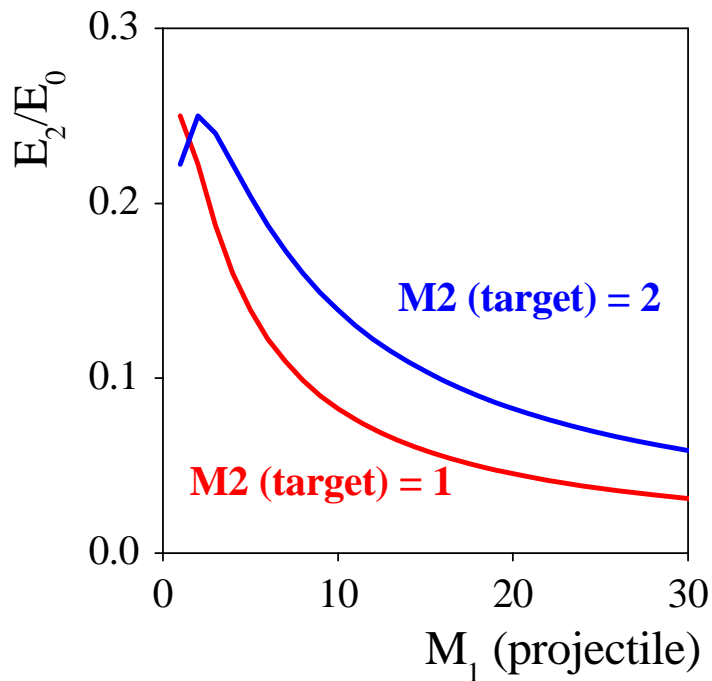
RBS spectrum of Ta implanted in Si



ERD – Elastic recoil detection

$$M \text{ (projectile)} > M \text{ (target)}$$

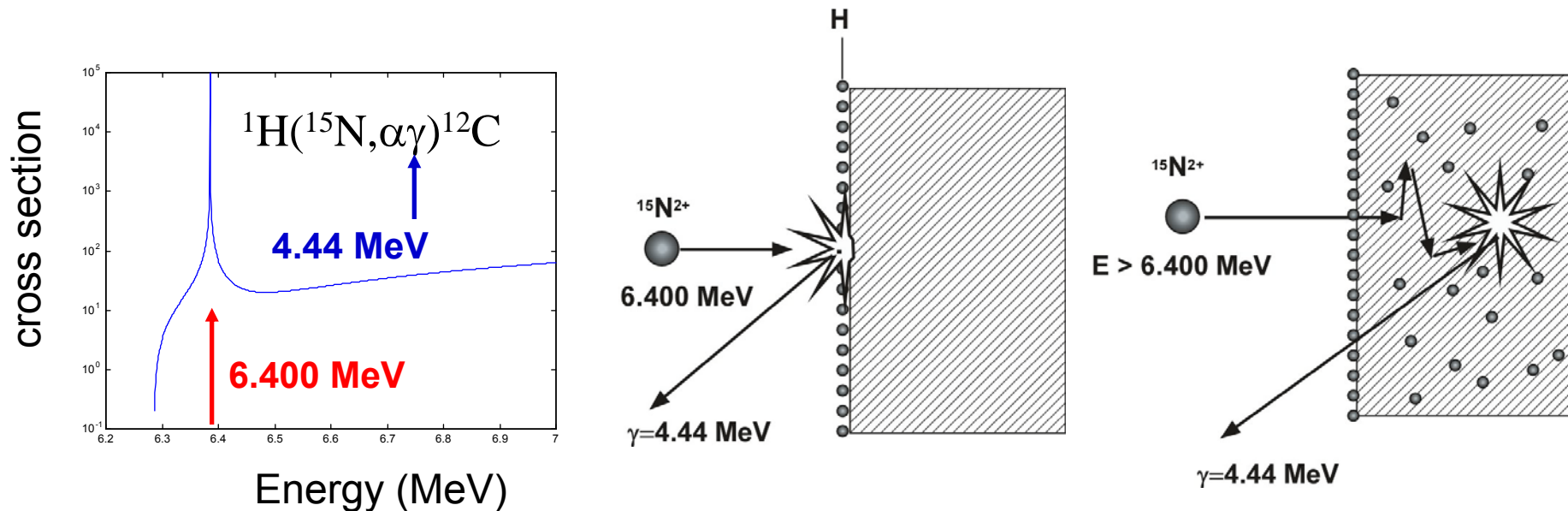
Analysis of the distribution of light elements (H, He, etc) in solids



Nuclear Resonant Reaction Analysis (NRRA)

Nuclear reactions with sharp resonances are used to study impurity concentration in solids

Example: The reaction ${}^1\text{H}({}^{15}\text{N}, \alpha\gamma){}^{12}\text{C}$ equiv. ${}^{15}\text{N}(\text{p}, \alpha\gamma){}^{12}\text{C}$



NRRA with the ${}^{15}\text{N}(\text{p}, \alpha\gamma){}^{12}\text{C}$ reaction measures the hydrogen concentration

Resonances used in NRRA

Table 15-1. Nuclear reactions used for the detection of charged particles produced in nuclear reactions (from Götz and Gärtner, 1988).

Nucleus	Reaction	Q-value Q [MeV]	Incident energy ¹⁾ E ₁ [MeV]	Emitted energy E' ₁ [MeV]	Cross-section (dσ/dΩ) _{NR} [mb/sr]	Mylar thickness Z _{sf} [μm]
² H	² H(d,p) ³ H	4.032	1.0	2.3	5.2	14
² H	² H(³ He,p) ⁴ He	18.352	0.7	13.0	61	6
³ He	³ He(d,p) ⁴ He	18.352	0.45	13.6	64	8
⁶ Li	⁶ Li(d,α) ⁴ He	22.374	0.7	9.7	6	8
⁷ Li	⁷ Li(p,α) ⁴ He	17.347	1.5	7.7	1.5	35
⁹ Be	⁹ Be(d,α) ⁷ Li	7.153	0.6 ²⁾	4.1	1	6
¹¹ B	¹¹ B(p,α) ⁸ Be	8.586	0.65	5.57(α ₀)	0.12(α ₀)	10
		5.65	0.65	3.70(α ₁)	90(α ₁)	10
¹² C	¹² C(d,p) ¹³ C	2.722	1.20	3.1	35	16
¹³ C	¹³ C(d,p) ¹⁴ C	5.951	0.64	5.8	0.4	0.6
¹⁴ N	¹⁴ N(d,α) ¹² C	13.574	1.5	9.9(α ₀)	0.6(α ₀)	23
		9.146	1.2	6.7(α ₁)	1.3(α ₁)	16
¹⁵ N	¹⁵ N(p,α) ¹² C	4.964	0.8 ³⁾	3.9	≈15	12
¹⁶ O	¹⁶ O(d,p) ¹⁷ O	1.917	0.90	2.4(p ₀) ⁴⁾	0.74(p ₀)	12
		1.05	0.90	1.6(p ₁)	4.5(p ₁)	12
¹⁸ O	¹⁸ O(p,α) ¹⁵ N	3.980	0.73 ³⁾	3.4	15	11
¹⁹ F	¹⁹ F(p,α) ¹⁶ O	8.114	1.25	6.9	0.5	25
²³ Na	²³ Na(p,α) ²⁰ Ne	2.379	0.592	2.238	4	6
³¹ P	³¹ P(p,α) ²⁸ Si	1.917	1.514	2.734	16	5)

¹⁾ For laboratory emission angle 150° with recoil nucleus in ground state (excited state)

²⁾ 0.6 MeV is optimum for Be in light Z matrix and 1.6 MeV for high Z matrix

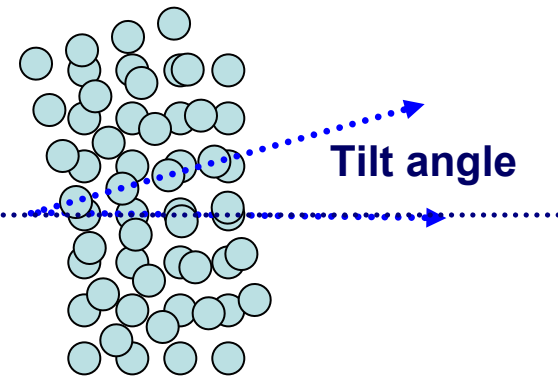
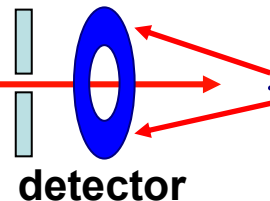
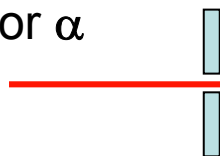
³⁾ Maximum energy for Mylar to stop backscattered proton

⁴⁾ Measured at θ=164°

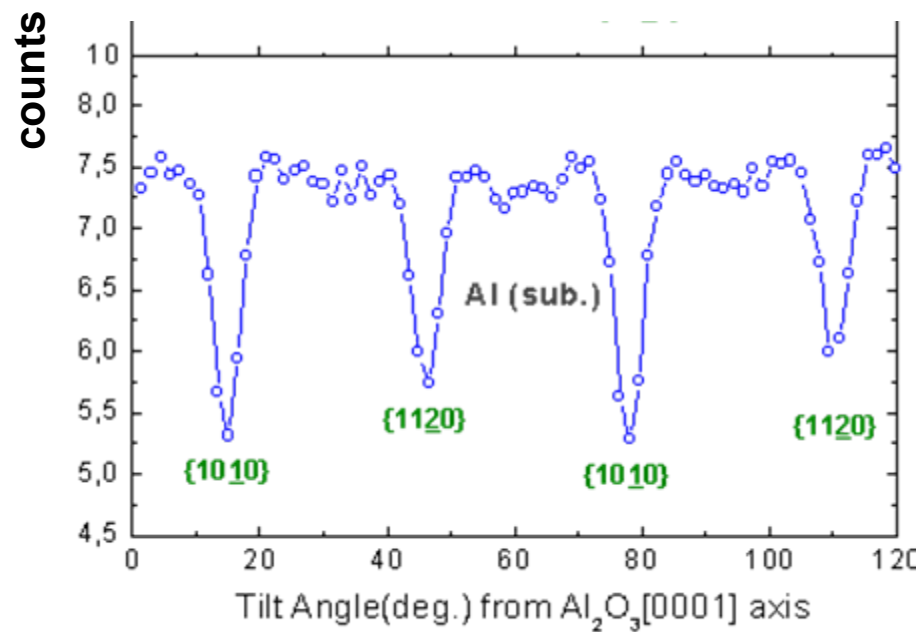
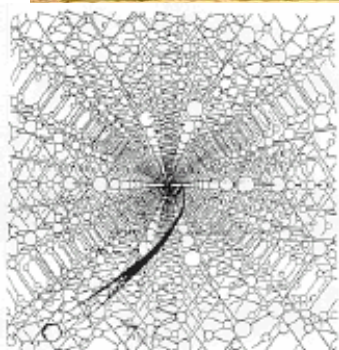
⁵⁾ range of α < range of proton

The Channeling Effect

Ion beam p or α



Single crystal
e.g. Al_2O_3



Emission channeling

Determination of the position of implanted ions

